VIVEKANANDA COLLEGE, ALIPURDUAR

Test Examination-2025 Subject: Chemistry Paper- DSC, 2nd Semester Assignment

F.M-30

A. Answer <u>any four</u> questions from the following: (5x4=20)

- 1. What is the significance of the negative sign in Bohr's equation relating the energy of H-atom? State the limitations of Rutherford's atomic model. (3+2)
- 2. Draw the shape of $dx^2-y^2 \ \& \ dz^2$ orbital's. Write the significance of spin quantum number.
- 3. The dipole moment of \mathbf{CO}_2 is zero, but that of water is 1.85D- explain. Why o-nitro phenol is volatile in nature but p-nitro phenol doesn't. (2+3)
- 4. Which is more covalent between CuCl₂ and ZnCl₂ and why? NaCl is soluble in water but not in
- 5. Why we get PCI₅ but not the PH₅? Write examples of two such compounds in which expansion of octet of central atom takes place. (3+2)

B. Answer <u>any one</u> question from the following: (10x1=10)

- 1. Explain the geometry of XeF4 & CIF3 in terms of VSEPR theory. (5+5)
- 2. Explain Fajan's rule with proper example. (10)
- 3. In 2s orbital the maximum number of electron is 2- justify. Determine the ionization energy of Hatom. (5+5)

VIVEKANANDA COLLEGE, ALIPURDUAR

Test Examination-2025 Subject: Chemistry Paper- IDC, 2nd Semester Assignment

F.M.: 20

Time: 1 hr

A. Answer any two questions: (5x2=10)

- 1. Write the names and symbols of some sunscreen substances. What colorants are commonly added to the lipsticks? (2+3)
- 2. What are sulfa drugs? Give an example. Describe the preparation procedure of sulfa drugs. (1+1+3)
- 3. Why are synthetic fuels used as alternative fuels? Write the preparation of butadiene from ethanol in petrochemical industry. (2+3)

B. Answer <u>any one</u> question from the following: (10x1=10)

- 1. Write four petrochemicals derived from hydrocarbons and their uses. (2.5x4=10)
- 2. Prepare Acyclovir from benzoyl chloride. (10)
- 3. Show the synthesis of Diazepam from p-nitrochlorobenzene.

VIVEKANANDA COLLEGE, ALIPURDUAR

4th SEMESTER

INTERNAL ASSESSMENT

SUBJECT - CHEMISTRY

QUESTION TYPE – ASSIGNMENT

QUESTION MODE – ONLINE

PAPER – MINOR

SESSION – 2024-2025

FULL MARKS - 20

INTRUCTIONS****

- Each question carries 4 marks each.
- Students are to submit the hard copy of the assignment to the teachers in person.
- A4 sized paper properly stapled and assembled in a stick file is only acceptable.
- The assignment must be submitted within the stipulated date as given in the college website.
- 1. What is Wurtz reaction? Can it be used to prepare alkane containing odd number of 'C' atoms. Give the preparation of ethane by Wurtz reaction.
- 2. What is Saytzeff and Hoffmann elimination? Give the points of differences.
- 3. What is Electrophilic aromatic substitution reaction (ArS_{ϵ}^{2}) ? Is Friedel Craft alkylation an ArS_{ϵ}^{2} reaction –Explain .
- 4. How does the electronegativity of the 'C' atom varies with bond order? Give the equation of the Allred –Rochow to calculate the electronegativity, specify each of the terms involved.
- 5. How does the Viscosity of the liquids and gases varies with rise in temperature.

VIVEKANANDA COLLEGE, ALIPURDUAR

4th SEMESTER

INTERNAL ASSESSMENT

SUBJECT - CHEMISTRY

QUESTION TYPE – ASSIGNMENT

QUESTION MODE – ONLINE

PAPER – MDC

SESSION – 2024-2025

FULL MARKS - 30

INTRUCTIONS****

- Each question carries 5 marks each.
- Students are to submit the hard copy of the assignment to the teachers in person.
- A4 sized paper properly stapled and assembled in a stick file is only acceptable.
- The assignment must be submitted within the stipulated date as given in the college website.
- 1. Write a note Green chemistry. Mention its principles.
- 2. What is a green solvent? Is water a green solvent? If, yes Explain.
- 3. What is Green energy? How come it is important in today's scenario Explain.
- 4. Write down an notable chemical reaction which follows the principles of green chemistry. Also highlight its mechanistic pathway as to how it follows the principles of green chemistry.
- 5. Give the green synthesis of a compostable plastic made from corn.
- 6. How CO₂ is replacing environmentally harmful solvents . Explain.

VIVEKANANDA COLLEGE, ALIPURDUAR
6th SEMESTER
INTERNAL ASSESSMENT
SUBJECT - CHEMISTRY
QUESTION TYPE – ASSIGNMENT
QUESTION MODE – ONLINE
PAPER – DSE
SESSION – 2024-2025

FULL MARKS - 20

INTRUCTIONS*****

- Each question carries 4 marks each.
- Students are to submitt the hard copy of the assignment to the teachers in person.
- A4 sized paper properly stapled and assembled in a stick file is only acceptable.
- The assignment must be submitted within the stipulated date as given in the college website.
- 1. Write a note on Geochemical cycle of Nitrogen .
- 2. What is Ozone depletion ? How CFCs are responsible for Ozone depletion . Give the mechanism in detail
- 3. What is Geothermal energy? Is it a conventional source of energy? If ,Yes. why? Name two more such types of energy as Geothermal energy.
- 4. What is Nuclear fusion and Nuclear fission? What is the source of energy in sun Explain.
- 5. Give the manufacturing process for H₂SO₄ acid.

VIVEKANANDA COLLEGE, ALIPURDUAR

Test Examination-2025 Subject: Chemistry Paper- SEC, 6th Semester

F.M.: 20

Time: 1 hr

A. Answer any two questions: (5x2=10)

- Name the chemicals used for vanishing cream. Write down the main ingredients of hair dye. (3+2)
- 2. Differentiate between cold cream and vanishing cream. What are the typical surfactants used in the preparation of shampoo? (2+3)
- 3. Why are 'Kaolin' and 'iron oxide' used in face powder? Name two essential oils used in cosmetics. (3+2)

B. Answer <u>any one</u> question from the following: (10x1= 10)

- 1. Write down the preparation of Nail Lacquers. Mention the ingredients used in shaving cream and describe its functioning. (5+5=10)
- 2. Write short note on- (2.5x4=10)
 - a) Anti-dandruff shampoo; b) Hair spray; (c) Artificial flavours; d) Cleaning action of shampoo.